

# SODIUM HYDROXIDE (NaOH) PRODUCT MATERIAL SAFETY DATA SHEET (MSDS)

## **REVISION HISTORY**

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Author:	On File	
Addition.	Gemma Rojas	Date
	Quality Control Supervisor	
Reviewed:	On File	
reviewed.	Esmeralda Santos	Date
	Head – Quality Control	
Reviewed:	On File	
	Alfredo M Reyes Jr VP Manufacturing	Date
	vr Mandiacturing	
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	natures on this page are applicable to all page	
Approved:		
	Document Controller	Date

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#### 1 DISCLAIMER

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## 2 SODIUM HYDROXIDE (NaOH)

#### 2.1 PRODUCT IDENTIFICATION

Chemical Name Sodium Hydroxide

Trade Name Caustic Soda, 50 %

Synonyms Liquid Caustic Soda, Caustic, Soda

Lye, Lye Solution

#### 2.2 HAZARDS IDENTIFICATION

This product may be corrosive, toxic and a major potential hazard upon contact to skin, eyes.

#### **TOXICITY ROUTES OF EXPOSURE**

Ingestion can cause severe burning and pain in lips, mouth, tongue, throat and stomach. Death can result from ingestion.

#### **OVEREXPOSURE**

- 1. Causes burns and scarring.
- 2. Can cause serious damage to all body tissues contacted

#### MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE:

Chronic eye or skin conditions

## 2.3 COMPOSITION / INGREDIENTS

**Sodium Hydroxide, %** 48-52% by weight

Chemical FormulaNaOHMolecular Weight40 g/moleCAS Registry No:1310-73-2

#### 2.4 FIRST AID MEASURES

Skin Remove contaminated clothing and immediately wash skin for a minimum of 15

minutes.

Eyes Immediately flush eyes with large amount of water, occasionally lifting the

upper and lower eyelids and rotating the eyeballs. Continue flushing for a

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minimum of 15 minutes

Inhalation Remove to fresh air. If breathing stops, administer artificial respiration

Ingestion DO NOT induce vomiting. If person is conscious, give 2 or more glasses of

water. If unconscious, never give anything by mouth

SEEK MEDICAL ATTENTION IMMEDIATELY

## 2.5 FIRE FIGHTING MEASURES

Autoignition Point Not Applicable. Flammability / Explosive limits: Not Applicable.

Fire / Explosion Hazards: Contact with strong acids may generate enough

heat to ignite combustibles.

Fire Prevention / Extinguishing Media Not Applicable.

## 2.6 <u>ACCIDENTAL RELEASE MEASURES</u>

**IN CASE OF SPILL OR RELEASE:** Completely contain spilled material with dikes, sandbags, etc., and prevent run off into the ground or surface waters or sewers. Recover as much caustic material as possible into containers for disposal. Add water and neutralize remaining caustic material with dilute hydrochloric acid, citric acid or another solid acidic material to a pH between 6 and 9. Collect neutralized caustic with a dry sorbent. Flush residual neutralized waste to the drain with excess water.

## 2.7 HANDLING AND STORAGE

Storage Requirements: Keep container tightly closed

For Small Volumes: Maybe stored in plastic jugs

For Large Volumes: Store in steel storage tanks

Incompatible Materials: Store away from acids (Refer to Section X)

## 2.8 EXPOSURE CONTROLS AND PROTECTION

Ventilation: 2 mg/m3

Protective Equipment for the Eyes and Goggles, respirator, disposable latex/ rubber apron,

Skin: PVC rain suit, rubber boots with pant legs over boots

Precautionary Hygiene Control Avoid contact with skin, eyes, and clothing. Do not

Measures: breathe mist or vapor. Wash thoroughly after

handling. Safety showers and eye wash fountains should be available in storage and handling area

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## 2.9 PHYSICAL AND CHEMICAL PROPERTIES

State liquid

Appearance colorless or slightly turbid

**Odor** Irritating

pH Strong base >14

**Boiling Point** 145 degrees Centigrade for 50%

NaOH solution

Flash Point Not determined

Specific Gravity 1.51-1.54

Vapor Pressure ~6.3mmHg at 40 degrees

Centigrade

Solubility In: WATER: miscible

ACID: miscible

### 2.10 STABILITY AND REACTIVITY

Stability Stable under normal handling conditions.

Hazardous polymerization will not occur.

Hazardous decomposition product None.

Materials and conditions to avoid Chlorinated hydrocarbons, acetaldehyde, acrolein,

(incompatibility) are aluminum, chlorine triflouride, hydroquinone, maleic

anhydride, and phosphorous pentoxide.

Dilution with water evolves large quantity of heat.

## 2.11 TOXICOLOGICAL INFORMATION

Effects from skin contact: Contact with skin can cause severe burns with deep

ulcerations. Contact with solution or mist can cause

multiple burns with temporary loss of hair at burn site.

Effects from eye contact Liquid in the eye can cause severe destruction and

blindness. These effects can occur rapidly affecting all

parts of the eye. Mist can cause

irritation with high concentration causing destructive burns.

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Cancer Information Not Applicable

#### 2.12 ECOLOGICAL INFORMATION

EcoToxicity Data: High basicity may pose potential hazard to plant and

marine life.

#### 2.13 <u>DISPOSAL CONSIDERATIONS</u>

Dispose of in accordance with all Government and Local regulations.

#### 2.14 TRANSPORT INFORMATION

<u>Transportation of Dangerous Goods</u>

TDG Classification Do not ship by air

DOT Hazard Classification Class 8; Corrosive: Group II
DOT Shipping Name Sodium Hydroxide ID: UN1824

## 2.15 REGULATORY INFORMATION

No data available.

## 2.16 OTHER INFORMATION

This MSDS contains information under the sixteen (16) section headings required by ISO 11014 "Safety Data Sheet for Chemical Products".

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